



Design of Ag–Ga–S_{2-x}Se_x-based eco-friendly core/shell quantum dots for narrow full-width at half-maximum using noble ZnGa₂S₄ shell material

Seung-Jae Lee¹ · Ji-Eun Lee² · Chang-Jin Lee¹ · Ui-Hyun Jeong¹ · Woo-Guk Lee³ · Hyo-Jun Kwon¹ · Min-Won Kim¹ · Tae-Hun Shim¹ · Jea-Gun Park^{1,3}

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Abstract

I–III–VI-based quantum dots (QDs) are promising eco-friendly light-emitting materials for next-generation displays. Thus far, I–III–VI-based QDs (i.e., AgIn_{1-x}Ga_xS₂) are still insufficient as light-emitting materials for display fields because of their deteriorated optical properties caused by lattice mismatch between the core/shell materials. We designed a novel passivation process using a shell material (i.e., ZnGa₂S₄) with crystalline structure identical to that of AgGaS_{2-x}Se_x core QDs (tetragonal). Green and red light-emitting AgGaS_{2-x}Se_x/ZnGa₂S₄ core/shell QDs demonstrated superior optical properties owing to the minimum lattice mismatch between the AgGaS_{2-x}Se_x core and ZnGa₂S₄ shell materials. In particular, a higher quantum yield (QY) and narrow full-width at half-maximum (FWHM) for the green and red light-emitting QDs were achieved, i.e., QYs of 69% and 64% and FWHMs of 26 and 28 nm, respectively. In addition, this remarkable enhancement of optical properties resulted in an increase of ~94.4% in the color-space agreement with ITU-R recommendation BT.2020. This means that AgGaS_{2-x}Se_x-based core/shell QDs have the potential to be used as eco-friendly QDs for light-emitting materials in next-generation displays.

Keywords AgGaS_{2-x}Se_x-based QD · FWHM · PL · Crystalline structure · Color gamut · BT 2020

1 Introduction

Quantum dots (QDs) have been researched intensively for several luminescent materials because of the adjustability of their properties, including wavelength, efficiency (i.e., high QY), stability, and color purity (i.e., narrow FWHM). For this reason, many studies have been conducted on the photovoltaic, detector, bio-imaging, and display fields in recent decades [1–4]. In particular, light-emitting materials for display applications (i.e., QD-OLED and QLED) were found to require more than 95% QY and less than 30 nm

narrow FWHM [5–7]. For instance, cadmium (Cd)-based QDs (CdSe QDs) of the II–VI materials have superior optical properties with high QY (over 95%), high stability, and narrow FWHM (> 25 nm). However, due to environmental issues, the use of Cd has been restricted in display applications according to the European Restriction of Hazardous Substances (RoHS) specification [8–11]. By contrast, light-emitting materials for display applications based on indium phosphide (InP) of III–V materials comply with RoHS standards. In particular, the InP-based QD of a III–V material structure (bulk bandgap energy of ~1.35 eV) can control the energy from near blue (~2.5 eV) to near infrared (~1.7 eV) by adjusting the core size [12–15]. Although eco-friendly InP-based QDs have been actively studied, they have limitations in improving the narrow FWHM (less than 30 nm) because of their large lattice covalency (the electrons shared with other atoms), resulting in the broad FWHM characteristics of InP-based QDs [16–18]. In general, the lattice covalency can be characterized using Phillips ionicity, which represents a quantified value for the type of chemical bonds between ionic (larger Phillips ionicity) and covalent

✉ Jea-Gun Park
parkjgl@hanyang.ac.kr

¹ Department of Electronic Engineering, Hanyang University, Seoul 04763, Republic of Korea

² Department of Information Display Engineering, Hanyang University, Seoul 04763, Republic of Korea

³ Department of Nanoscale Semiconductor Engineering, Hanyang University, Seoul 04763, Republic of Korea

bonding (smaller Phillips ionicity) [19–21]. Note that the Phillips ionicity of InP-based QDs is small because of the covalent bonding between In^{3+} and P^{3-} atoms, and thus they have a relatively larger lattice covalency compared to CdSe-based QDs (Phillips ionicities of InP and CdSe are 0.421 and 0.699, respectively) [17, 19]. Accordingly, the larger lattice covalency of InP-based QDs intrinsically requires a high growth temperature ($\sim 300^\circ\text{C}$) and reactive precursors (Tris(trimethylsilyl) phosphide). Therefore, InP-based QDs have the optical properties of broad particle size distribution (broad FWHM), because they require high energy for the nucleation and growth steps of the InP core QDs [22–24]. Thus far, it has been reported that $\text{AgIn}_x\text{Ga}_{1-x}\text{S}_2$ ($\text{AgIn}_x\text{Ga}_{1-x}\text{S}_2/\text{Ga}_2\text{S}_3$)-based core/shell QDs of I–III–V core materials have high Phillips ionicities (~ 0.813) [19, 25]. However, the core/shell QDs of $\text{AgIn}_x\text{Ga}_{1-x}\text{S}_2/\text{Ga}_2\text{S}_3$ are difficult to passivate perfectly with a shell material because of the lattice mismatch ($\sim 8.5\%$) of crystalline structures between the $\text{AgIn}_x\text{Ga}_{1-x}\text{S}_2$ core (tetragonal) and Ga_2S_3 shell (hexagonal), resulting in poor optical properties of QY (less than 50%) and wider FWHM (wider than 40 nm) [26–30].

In this study, $\text{AgGaS}_{2-x}\text{Se}_x$ -based core QDs as light-emitting materials with ionic bonds of larger phillip ionicity (~ 0.729) were designed for green and red light-emitting core QDs. In addition, shell structures (Ga_2S_3 , ZnGa_2S_4 , and ZnS) were introduced to achieve the optical properties of

high QY and narrow FWHM of the $\text{AgGaS}_{2-x}\text{Se}_x$ -based core/shell QDs, as shown in Fig. 1. Furthermore, we conducted a novel synthesis of I–III–VI-based core/shell QDs with a very narrow FWHM characteristic (less 30 nm) through passivation using a shell material (ZnGa_2S_4) with a crystalline structure identical to that of $\text{AgGaS}_{2-x}\text{Se}_x$ core QDs (tetragonal) to minimize lattice mismatch ($\sim 6.5\%$) between core/shell materials. In particular, the optical properties, crystallinities, and morphologies of the three types of shell structures on the $\text{AgGaS}_{1-x}\text{Se}_x$ -based core QDs were investigated to enhance the color-space matching with the BT 2020 standard by improving the emission wavelength and FWHM.

2 Experiments and discussion

2.1 Materials

Oleylamine (OAm, 70%), 1-dodecanethiol (DDT, > 98%), silver acetate ($\text{Ag}(\text{OAc})$, 99.99%), gallium(III) acetylacetonate ($\text{Ga}(\text{acac})_3$, 99.99%), tetramethylthiuram disulfide (TMTDS, 97%), zinc acetate ($\text{Zn}(\text{OAc})_2$, 99.99%), trioctylphosphine (TOP, 97%), selenium (trace metals basis, 100 mesh, and 99.99%), and oleic acid (OA, technical grade 90%) supplied by Sigma-Aldrich were used.

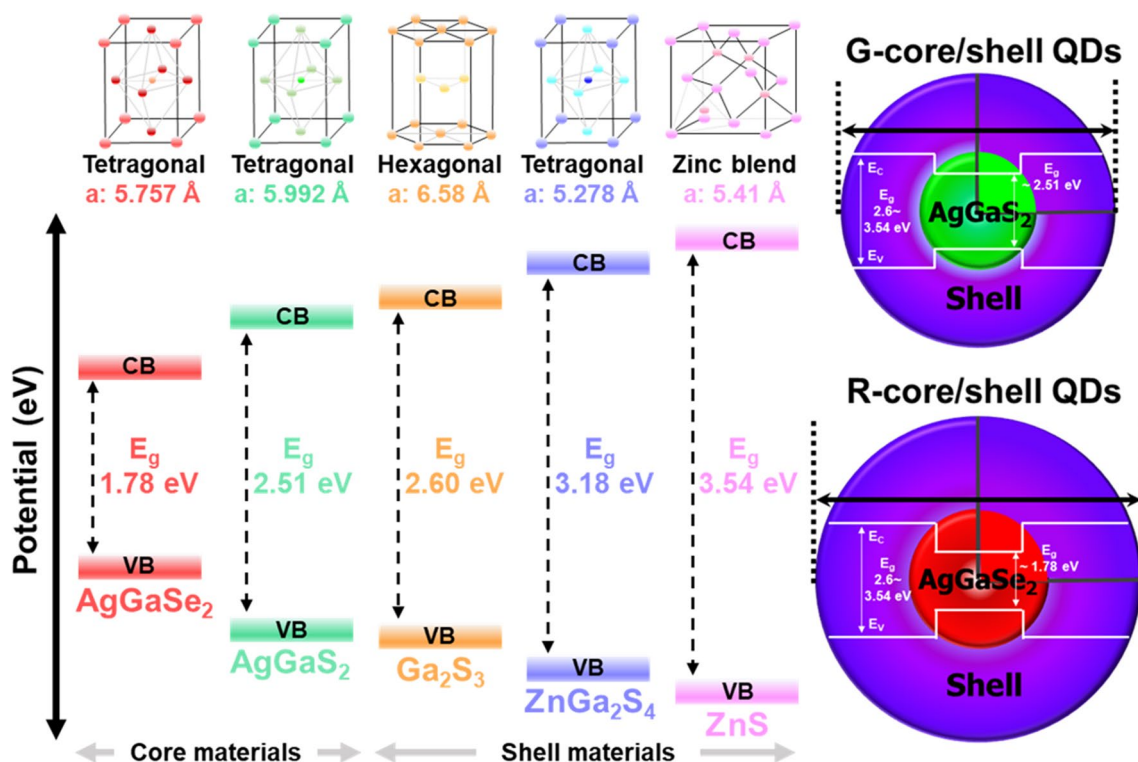


Fig. 1 Crystalline structure and energy bandgap diagram according to core/shell QDs. The schematic of a core/shell QD structure depending on the core materials (right images)

2.2 Preparation of precursors

Ag (OAc)-Oam (0.5 M) was prepared by dissolving 1 mmol of Ag(Oac) powder in 2 ml of OAm in an N₂-filled glove box. A 0.5 M TMTDS-OAm precursor was prepared with 1 mmol of TMTDS dissolved in 2 ml of OAm in an N₂-filled glove box. A 0.5 M Se-OAm precursor was prepared with 1 mmol of Se powder dissolved in 2 ml of OAm in an N₂-filled glove box. Then, all the precursors were evacuated at 150 °C for 1 h and heated to 170 °C under N₂ flow. A 0.5 M Ga(acac)₃-ODE precursor was prepared with 1 mmol of Ga(acac)₃ dissolved in 2 ml of ODE in an N₂-filled glove box. A 1 M Zn(oleate)₂ precursor was prepared by dissolving 5 mmol of Zn(OAc)₂ in 10 ml of ODE and oleic acid, and the precursors were evacuated at 150 °C for 1 h and heated to 250 °C under N₂ flow.

2.3 Synthesis of green and red light-emitting AgGaS₂ and AgGaSe₂ core/shell QDs

For the AgGaS₂ and AgGaSe₂ for green and red light-emitting cores, 0.5 mmol of Ag(OAc) and Ga(acac)₃ each were loaded into a 100 mL three-neck flask with 14 mL OAm at room temperature. The mixture was heated to 150 °C while being stirred, and then degassed under a pressure of 100 mTorr for 1 sh. Then, the precursor solution for 0.6 mL of 0.5 M TMTDS-OAm (green-light emitting core QDs) or 0.4 mL Se-OAm (red light-emitting core QDs) was rapidly injected in the mixture at 200 or 220 °C for 40 min under N₂ flow. For ZnGa₂S₄ shell growth, 7.6 mL of Zn(oleate)₂, Ga(acac)₃, DDT, and ODE precursors were slowly added dropwise into the core mixture at 240 °C for 90 min. After the ZnGa₂S₄ shell growth, the solution was cooled to room temperature and then centrifuged with an excessive amount of acetone (at least 3 times) to eliminate the impurities of unreacted starting materials and side products. Finally, the precipitated QDs were redispersed in 5 mL of an organic solvent. The detailed morphology, crystallinity, crystalline structure, size distribution, and core/shell structures of the AgGaS_{1-x}Se_x-based core/shell QDs were characterized by

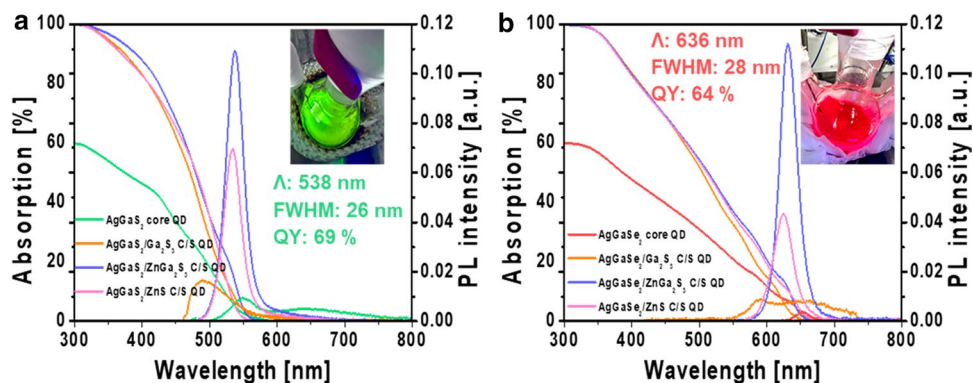
transmission electron microscopy (TEM) at an acceleration voltage of 200 kV, energy-dispersive X-ray spectroscopy (EDX), and X-ray diffraction (XRD). The optical properties were measured by UV–visible absorption spectra using a Carry-5000 (Varian), and the photoluminescence (PL) spectra were measured using a HeCd laser source at an 325 nm excitation wavelength. The FWHM and absolute PL QY were measured using a QE-2100 (Otsuka Electronics).

3 Results and discussion

To estimate the optical properties depending on the crystalline structures of the shell materials, three types of shell materials (Ga₂S₃, ZnGa₂S₄, and ZnS) on AgGaS_{1-x}Se_x core QDs were designed, as shown in Fig. 1. In general, the lattice mismatch between the core and shell induces strain at their interface, leading to deteriorating optical properties. Thus, to improve the optical properties, it is necessary to apply an identical crystalline structure of the core with shell materials, for a minimum lattice mismatch between the core and shell, and wider bandgap energy of the shell materials than of the core material.

As shown in Fig. 2, it was confirmed that the optical properties (emitting wavelength, QY, and FWHM) depend on the shell bandgap energy and crystalline structures on the core, i.e., bulk bandgap energies of the AgGaSe₂, AgGaS₂, Ga₂S₃, ZnGa₂S₄, and ZnS materials were 1.78, 2.51, 2.60, 3.18, and 3.54 eV, respectively. The absorption and emission spectra of the G-AgGaS₂- and R-AgGaSe₂-based core/shell QD are presented in Fig. 2a, b, respectively. The absorption and PL spectra of the three types of shells passivated on the G-AgGaS₂- and R-AgGaSe₂-core QDs were obviously different. Accordingly, in the case of G-AgGaS₂-based core/shell QDs, the absorption spectra increased from the UV to the visible light region (300,550 nm) when passivated by shell materials (Ga₂S₃, ZnGa₂S₄, and ZnS). As a result, the QYs of three types of G-AgGaS₂-based core/shell QDs were 22%, 69%, and 52%, respectively, which translated to ~2.4, 7.6, and 5.7 times higher than those for the G-AgGaS₂ core

Fig. 2 Absorption and PL spectra of the green light-emitting AgGaS₂-based core/shell QDs and the red light-emitting AgGaSe₂-based core/shell QDs, **a** depending on the shell structures excited under 450 nm. Photographs (insets in **a** and **b**) of solution-based QDs under 365 nm hand-UV lamp for G-AgGaS₂/ZnGa₂S₄ and R-AgGaSe₂/ZnGa₂S₄ core/shell QDs, respectively



QDs (9%), respectively, as shown in Fig. 2a and Table 1. In addition, the FWHM of the two types (ZnGa_2S_4 and ZnS) of G-AgGaSe_2 -based core/shell QDs were 26 and 31 nm, respectively, which translated to ~ 0.62 and 0.74 times narrower than those of the G-AgGaSe_2 core QDs (42 nm), respectively, because the shell growth of ZnGa_2S_4 or ZnS was dominantly reacted rather than diffusion effect of Ga^{3+} and Zn^{2+} ions, as shown in Fig. 2a and Table 1. In the case of shell passivation with Ga_2S_3 , the FWHM of the $\text{G-AgGaSe}_2/\text{Ga}_2\text{S}_3$ core/shell QDs (55 nm) was wider than that of the G-AgGaSe_2 core QDs (42 nm), since the diffusion effect of Ga^{3+} ions was mainly reacted rather than Ga_2S_3 shell growth [31]. Thus, the QY and FWHM of the $\text{G-AgGaSe}_2/\text{ZnGa}_2\text{S}_4$ core/shell QDs with an identical crystalline structure (tetragonal) among three types of shell materials were 69% and 26 nm, respectively, which translated to an enhancement of ~ 7.6 and 0.62 times compared with those for the AgGaSe_2 core QDs (9% and 42 nm), respectively.

For R-AgGaSe_2 -based core/shell QDs, the absorption spectra increased from the UV- to visible-light region (300–650 nm) when passivated by shell materials. As a result, the QYs of three types of R-AgGaSe_2 -based core/shell QDs were 19%, 64%, and 46%, respectively, translating to ~ 3.1 , 10.6 , and 7.6 times higher than those for the R-AgGaSe_2 core QDs (6%), respectively, as shown in Fig. 2b and Table 1. In addition, the FWHM of two types (ZnGa_2S_4 and ZnS) of R-AgGaSe_2 -based core/shell QD were 28 and 34 nm, respectively, which translates to ~ 0.75 and 0.91 times narrower than those for the R-AgGaSe_2 core QDs (37 nm), as shown in Fig. 2b and Table 1. The FWHM of the $\text{R-AgGaSe}_2/\text{Ga}_2\text{S}_3$ core/shell QDs, in the case of the shell passivated with Ga_2S_3 material, the FWHM of the $\text{R-AgGaSe}_2/\text{Ga}_2\text{S}_3$ core/shell QDs (141 nm) is much wider than that of the R-AgGaSe_2 core QDs (37 nm), because not only Ga^{3+} cations but also S^{2-} anion atoms diffuse into the

core. Therefore, the Ga_2S_3 shell was unstably passivated on the R-AgGaSe_2 core QDs. Therefore, the QY and FWHM of the $\text{R-AgGaSe}_2/\text{ZnGa}_2\text{S}_4$ core/shell QDs among three types of shell materials were 64% and 28 nm, respectively, which translates to ~ 10.6 and 0.75 times higher than those for the AgGaSe_2 core QDs (6% and 7 nm), respectively. Thus, because the lattice mismatch between the AgGaSe_2 core and ZnGa_2S_4 shell ($\sim 6.5\%$) was smaller than that of two types of $\text{AgGaSe}_2/\text{Ga}_2\text{S}_3$ ($\sim 8.5\%$) and $\text{AgGaSe}_2/\text{ZnS}$ ($\sim 7.7\%$) core/shell QDs, the optical properties of the $\text{AgGaSe}_2/\text{ZnGa}_2\text{S}_4$ core/shell structure exhibited the best results. Thus, we demonstrate an improvement in the optical properties as a result of the minimal lattice mismatch between the core and shell materials. To confirm this, we discuss it later through XRD and HR-TEM analyses.

To understand why the optical properties of the ZnGa_2S_4 shell material were remarkably higher than those of the Ga_2S_3 and ZnS shell materials, the crystallinities of the three types of shell structures were confirmed using selected area diffraction (SAED) and XRD patterns, as shown in Fig. 3a, b. The XRD peak positions of the diffraction peaks ((112), (204), (220), (312), and (224)) of the tetragonal crystalline structure were affected by the differences in the three types of crystalline structures (shell materials). Note that the distinct rings originating from (112), (204), (220), (312), and (224) in the SAED patterns correlated well with the dominant XRD diffraction peaks for the core and three types of core/shell structures. However, in the case of the $\text{AgGaSe}_2/\text{Ga}_2\text{S}_3$ and $\text{AgGaSe}_2/\text{ZnS}$ core/shell structures, the XRD peak positions were affected by the crystalline structure of the hexagonal and zinc blends, resulting in a shift in the XRD peak position. The highest crystallinity shown from the relative FWHM of the three types of crystalline structures for the shell materials was observed in the same tetragonal crystalline structure between the core

Table 1 Optical properties of the green and red light-emitting AgGaSe_2 -based core/shell QDs depending on the shell structures

Emitting light color	Core/shell structure	PL λ_{max} ^a [nm]	Abs. λ_{max} ^b [nm]	Stokes shift ^c [nm]	FWHM [nm]	PL-QY ^d [%]
Green light-emitting core/shell QDs	AgGaSe_2 core	549	531	18	42	9
	$\text{AgGaSe}_2/\text{Ga}_2\text{S}_3$ C/S	491	482	9	55	22
	$\text{AgGaSe}_2/\text{ZnGa}_2\text{S}_4$ C/S	538	515	23	26	69
	$\text{AgGaSe}_2/\text{ZnS}$ C/S	534	518	16	31	52
Red light-emitting core/shell QDs	AgGaSe_2 core	651	636	15	37	6
	$\text{AgGaSe}_2/\text{Ga}_2\text{S}_3$ C/S	592	586	6	141	19
	$\text{AgGaSe}_2/\text{ZnGa}_2\text{S}_4$ C/S	631	615	16	28	64
	$\text{AgGaSe}_2/\text{ZnS}$ C/S	624	610	14	34	46

^aWavelength of photoluminescence peak

^bFirst peak wavelength of absorption spectra

^cDifference (in wavelength) between abs. λ_{max} and PL λ_{max}

^dMeasured by absolute PL QY using QE-2100 (Otsuka Electronics)

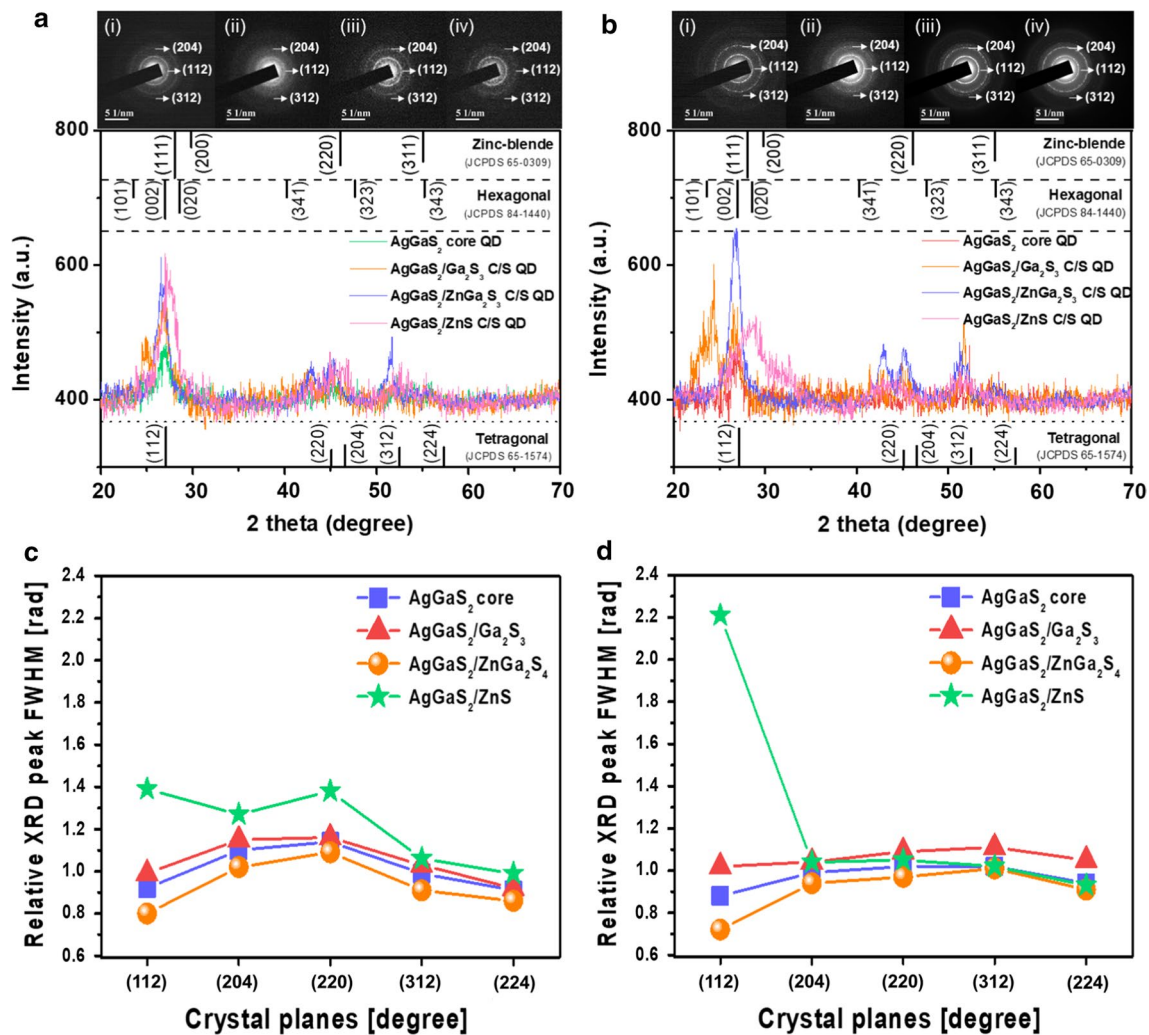


Fig. 3 X-ray diffraction (XRD) patterns of the AgGaS₂- **a** and AgGaS₂-based core/shell QDs **b** depending on the shell structures by the Crystalline structure, and corresponding selected area electron

diffraction (SAED) patterns (above image). Dependency of the relative XRD peak FWHM for tetragonal crystal planes. The AgGaS₂- **(c)** and AgGaS₂-based core/shell QDs **d**, respectively

(G-AgGaS₂ and R-AgGaS₂) and shell (ZnGa₂S₄) materials. As a result, the relative XRD FWHM at the peak position of the (112) plane of G-AgGaS₂/Ga₂S₃ (0.99 rad. a.u.), G-AgGaS₂/ZnGa₂S₄ (0.8 rad.), and G-AgGaS₂/ZnS core/shell QDs (1.39 rad.) were ~ 1.07, 0.87, and 2.1 times wider than that for the AgGaS₂ core QDs (0.92 rad.) eV, respectively, as shown in Fig. 3c. For the R-AgGaS₂-based core/shell QDs, the relative XRD FWHM at the peak position of the (112) plane of R-AgGaS₂/Ga₂S₃ (1.02 rad.), R-AgGaS₂/ZnGa₂S₄ (0.72 rad.) and R-AgGaS₂/ZnS core/shell QDs (2.21 rad.) were ~ 1.16, 0.81, and 2.51 times wider than that of the R-AgGaS₂ core QDs (0.88 rad.), respectively, as shown in Fig. 3d. In general, for XRD analysis, the crystallinity showing a narrower FWHM at the XRD peak position is better. Thus, from the results of the XRD analysis, the shell passivation with ZnGa₂S₄ material showed the best

crystallinity among the three types of shell materials, which resulted in perfect shell passivation on the core QDs.

The spherical-like shape, high crystallinity, good dispersity, and interplanar distance of the core and three types of core/shell structures were compared in detail using HR-TEM images, as shown in Fig. 4. For the G-AgGaS₂-based core/shell QDs, when the diameter of the core QDs was 4.1 ± 0.54 , the diameters of the core/shell structures passivated with Ga₂S₃, ZnGa₂S₄, and ZnS shell materials were 4.4 ± 0.51 , 6.8 ± 0.41 , and 6.5 ± 0.99 nm, respectively, as shown in Fig. 4a, d. In addition, for the R-AgGaS₂-based core/shell QDs, when the diameter of the core QDs was 6.7 ± 0.84 , the diameters of the core QDs and core/shell structures with Ga₂S₃, ZnGa₂S₄, and ZnS shell materials were 7.0 ± 0.92 , 9.6 ± 0.53 , and 8.9 ± 0.92 nm, respectively, as shown in Fig. 4e, h. When the G-AgGaS₂ and R-AgGaS₂-based core QDs were passivated with a Ga₂S₃

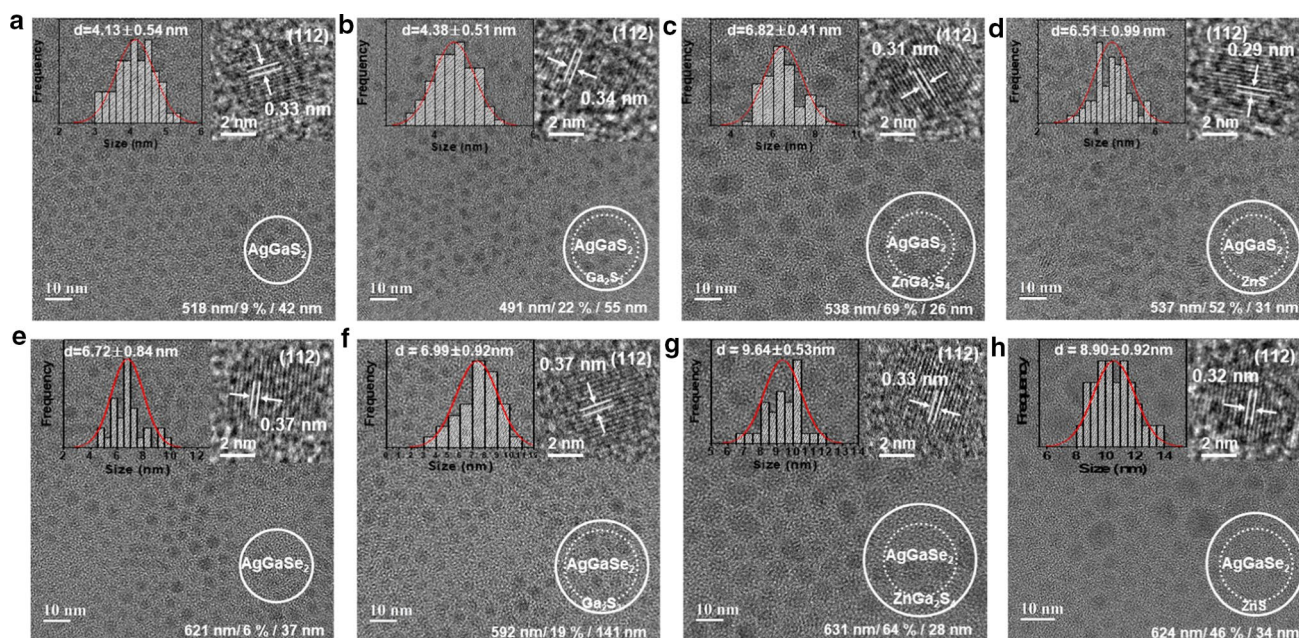


Fig. 4 High-resolution transmission electron microscopy (HRTEM) images of the green **a–d** and red light-emitting QDs **e–h** depending on the shell structures with a scale bar of 10 nm, size distribution (inset, upper left), and d-spacing values with a scale bar of 2 nm (inset, upper right)

shell material, it was confirmed that the Ga₂S₃ material was not grown as a shell layer and defect sites inside or on the core were passivated with Ga³⁺ ions, as shown in Fig. 4b, f [26, 27]. Furthermore, the interplanar distances of the G-AgGaS₂-based core QDs and R-AgGaSe₂-based core were 0.33 and 0.37 nm at the (112) plane, as shown in the HR-TEM images of the insets in Fig. 4a, e. Moreover, the interplanar distances of the core/shell structures for the three types of shell materials were altered by the difference in the lattice constant depending on the shell materials. From the XRD patterns and TEM images, it was confirmed that the perfectly spherical shape, thickest shell growth, and narrowest FWHM of G-AgGaS₂ (26 nm) and R-AgGaSe₂-based core with ZnGa₂S₄ shell QDs (28 nm) were attributed to the lattice mismatch of ZnGa₂S₄-based core/shell QDs (~6.5%) being the smallest among Ga₂S₃- and ZnS-based core/shell QDs (~8.5% and 7.7%).

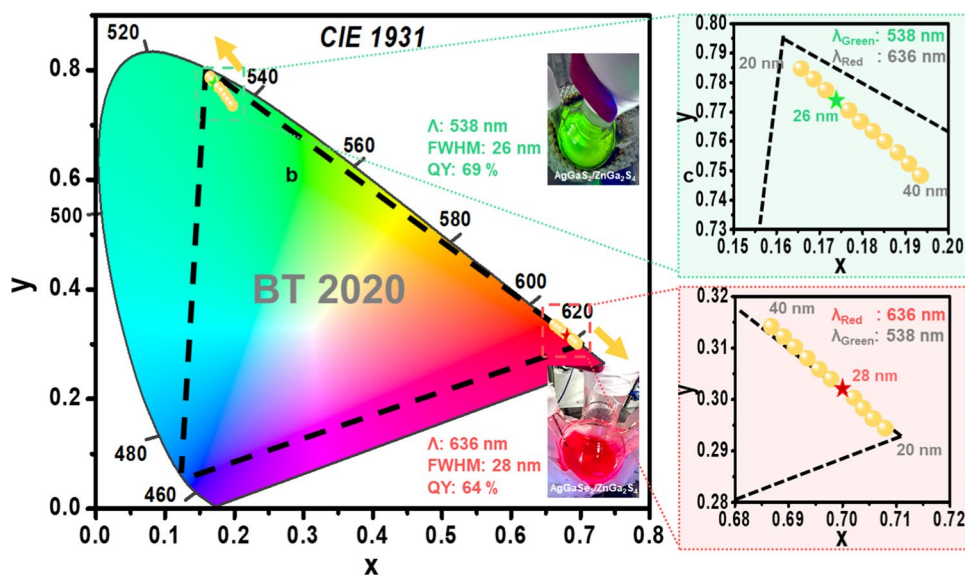
Note that the narrow FWHM of the PL spectra of QDs is an important parameter of color purity (color reproducibility) for display characteristics. In addition, QLED–LCD and QD–OLED displays, which use red, green, and blue (RGB) colors transmitted through respective RGB color filters (CFs) from a backlight unit, should comply with the BT 2020 color standard, which is a set of specifications for various aspects of video broadcasting recommended by the ITU as standards for ultra-high-definition (UHD). Therefore, we simulated the agreement with the BT 2020 color standard, according to the FWHM of the G-AgGaS₂/ZnGa₂S₄ and R-AgGaSe₂/ZnGa₂S₄ core/shell QDs without CFs at

a fixed blue wavelength (455 nm) and FWHM (20 nm), extracted from the green and red light PL spectra, as shown in Fig. 5. A decrease in the FWHM of the green and red light PL spectrum to 20 nm, indicated an improvement of 95.7% with the BT2020 agreement. Finally, in the case of the G-AgGaS₂/ZnGa₂S₄ and R-AgGaSe₂/ZnGa₂S₄ core/shell QDs, an improvement of 94.4% with the BT2020 agreement was achieved via optimal optical properties of the green (538 nm of the wavelength and 26 nm of the FWHM) and red light spectra (636 nm of the wavelength and 28 nm of the FWHM), as shown in the enlarged graph of Fig. 5.

4 Conclusions

QDs have been intensively researched as light-emitting materials for display applications because of their adjustable properties, such as the wavelength, efficiency (high QY), stability, and color purity (narrow FWHM of PL spectra). In particular, I–III–VI-based QDs (AgGaS_{2-x}Se_x) have attracted interest as promising eco-friendly (cadmium-free) light-emitting materials for next-generation displays. In addition, light-emitting materials for display applications (QD–OLED and QLED) should demonstrate the optical properties of the PL spectra FWHM of less than 30 nm for green and red light-emitting QDs. Although AgIn_{1-x}Ga_xS₂-based core/shell QDs (I–III–VI-based QDs) have been reported, they exhibit optical properties with an FWHM of ~40 nm. However, our I–III–VI-based QDs of AgGaS_{2-x}Se_x/ZnGa₂S₄

Fig. 5 Shift of coordinates in the color triangle space depending on the FWHM (from 40 to 20 nm) of the emitting spectra for the center peak points at 538 and 636 nm. Enlarged graphs for each region in green (top) and red (bottom) are shown



with a minimum lattice mismatch between the AgGaS_{2-x}Se_x core and ZnGa₂S₄ shell and an identical crystal structure (tetragonal) resulted in much narrower FWHMs of the green and red light-emitting QDs (26 and 28 nm). In conclusion, these results indicate that the AgGaS_{2-x}Se_x/ZnGa₂S₄ core/shell QDs are nearly comparable to the cadmium-based core/shell QDs (with less than 25 nm of FWHM). In addition, BT 2020 agreement between G-AgGaS_{2-x}Se_x/ZnGa₂S₄ and R-AgGaSe₂/ZnGa₂S₄ core/shell QDs was ~94.4% without the CFs.

Fundamentally, although AgIn_{1-x}Ga_xS₂-based core/shell QDs showed low FWHM (~28 nm), the QY of the green and red-light QDs was low (~70%), because they were passivated with a monolayer shell structure of ZnGa₂S₄. Therefore, further research to achieve a high QY using multi-layered shell structures with the same crystalline structure on AgIn_{1-x}Ga_xS₂-based core/shell QDs is necessary.

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References

1. Coe-Sullivan et al., ECS J. Solid State Sci. Technol. **2**, 3026–3030 (2013)
2. H. Chen et al., Opt. Express **25**, 102–111 (2017)
3. Luo, Z. et al., 2014 Proc. of SPIE 9005, 90050G1–7
4. M.A. Cotta, ACS Appl. Nano Mater. **3**, 4920–4924 (2020)
5. K.Y. Lai et al., Nanomaterials **12**, 2683 (2022)
6. M.K. Choi et al., npj Flex. Electron. **2**, 1 (2018)
7. J.R. Manders et al., Quantum dots for displays and solid state lighting, in *In Materials for Solid State Lighting and Displays*, ed. by A. Kitai (Wiley Series, Hoboken, 2016), pp.31–90
8. S.A. Kusuma et al., J. Clean. Prod. **366**, 132760 (2022)
9. S. Tsukuda et al., J Soc Inf Disp. **28**, 680–690 (2020)
10. H.S. Moon et al., Adv. Mater. **31**, 157–160 (2019)
11. A. Saha et al., Chem. Mater. **32**, 2148–2155 (2020)
12. D.A. Taylor et al., Chem. Mater. **33**, 4399–4407 (2021)
13. R. Toufanian et al., Front. Chem. **6**, 567–573 (2018)
14. Yi. Zhang et al., Prog. Photovoltaics Res. Appl. **6**, 571–596 (2022)
15. H. Meddeb et al., Adv. Energy Mater. **12**, 2200713 (2022)
16. X. Jiang et al., Micromachines **13**, 709 (2022)
17. E.J. Jang et al., ACS Energy Lett. **5**, 1316–1327 (2020)
18. E.M. Janke, J. Am. Chem. Soc. **140**, 15791–15803 (2018)
19. O.P. Singh et al., Phys. stat. sol. **137**, 97 (1986)
20. K. Hubner et al., Phys. stat. sol. **77**, 473 (1976)
21. J.R. Manders et al., in *Semiconductor Materials: an Introduction to Basic Principles*, ed. by B.G. Yacobi (Springer, Boston, 2003), pp.5–32
22. Y.J. Kwon et al., NPG Asia Mater. **13**, 13–37 (2021)
23. D.C. Gary et al., Chem. Mater. **27**, 1432–1441 (2015)
24. A.L. Efros et al., ACS Nano **15**, 6192–6210 (2021)
25. Y.S. Kim et al., Chem. Sci. **11**, 913–922 (2019)
26. T. Uematsu et al., NPG Asia Mater. **10**, 713–726 (2018)
27. T. Kameyama et al., ACS Appl. Mater. Interfaces **10**, 42844–42855 (2018)
28. W. Hoisang et al., Nanoscale Adv. **4**(2), 849–857 (2022)
29. A. Hirase et al., Phys. Chem. Lett. **11**, 3969–3974 (2020)
30. W. Hoisang et al., Nanomaterials **9**, 1769 (2019)
31. X. Peng et al., J. Am. Chem. Soc. **119**, 7019–7028 (1997)

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